Class Test-1, 2024-2025

Semester-4, DSCC-5, Organic Chemistry

FM: 20, Time: 48 minutes

Date of Examination: 18. 07. 25

Topic: Stereochemistry

Attempt any Four questions: 4 x 5

- 1. (a) Draw the potential energy diagram of *n*-butane for rotation around C2-C3 bond showing the conformers. Explain the relative stabilities of the conformer.
 - (b) Draw the most stable conformations of 1,2-dibromoethane and ethylene glycol and justify your answer. 2

3

3

2

3

3

2

- 2. (a) Draw the conformers of 1-chlorobutane for rotation about C1-C2 bond. Comment on the relative stabilities of the conformers.
 - (b) The rotational energy barriers in CH₃CH₂X (X = F, Cl, Br, I) are remarkably similar in magnitude despite considerable differences in the size of halogens. How do you explain for the fact.
- 3. (a) Explain the nature of stereoisomerism exhibited by the compounds of the formula $XYC=(C=)_nXY$ where n=1 and 2.
 - (b) For each of the following molecule, designate the R/S configuration after determining the priority order.

(a)
$$C = C = C \cup CI$$

 Et

(bi) CI
 $MeO CH_2OH$

- 4. (a) Explain the stereoisomerism of 6,6'-dinitrodiphenic acid and draw the energy profile diagram for racemization of its enantiomers on heating. Label each maximum and minimum with appropriate rotamers.
 - (b) Define atropisomerism with suitable examples.
- 5. (a) Find out the topicity relationship between the marked ligands in each of the following:

(a)
$$H_a$$
 H_b H_b H_a H_b H_b H_b H_b H_b H_b H_b

- (b) All stereogenic centres are not chiral. Justify with suitable examples
- 6. (a) Identify, with explanation, among the marked hydrogens the *pro-R* and *pro-S* ones in each case.

(b) Determine the configuration of the final product resulting from attack of MeMgBr onto the *Re*-face of *R*-2-hydroxypropanal and subsequent hydrolysis.

_____ End ____