

Organic Chemistry-4
Semester-4, CBCS
Course: CEMA CC-4-8-TH

Course taught by: Kaushik Basu, Department of Chemistry, SPCMC, Kolkata
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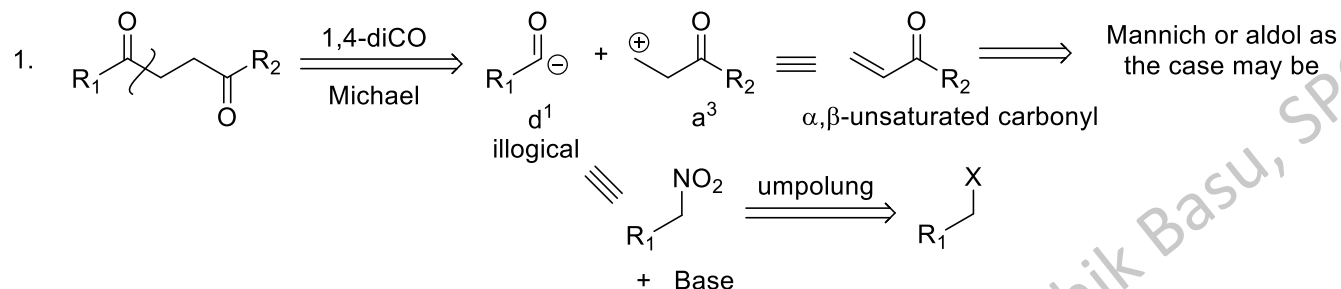
Recommended texts:

1. Study Guide to Organic Chemistry, Volume 2, by Saha, Chakraborty, Saha & Basu, Techno World, ISBN 9788192669588,
2. Organic Chemistry, Second Ed. by Clayden, Greeves & Warren, OUP, ISBN 9780198728719

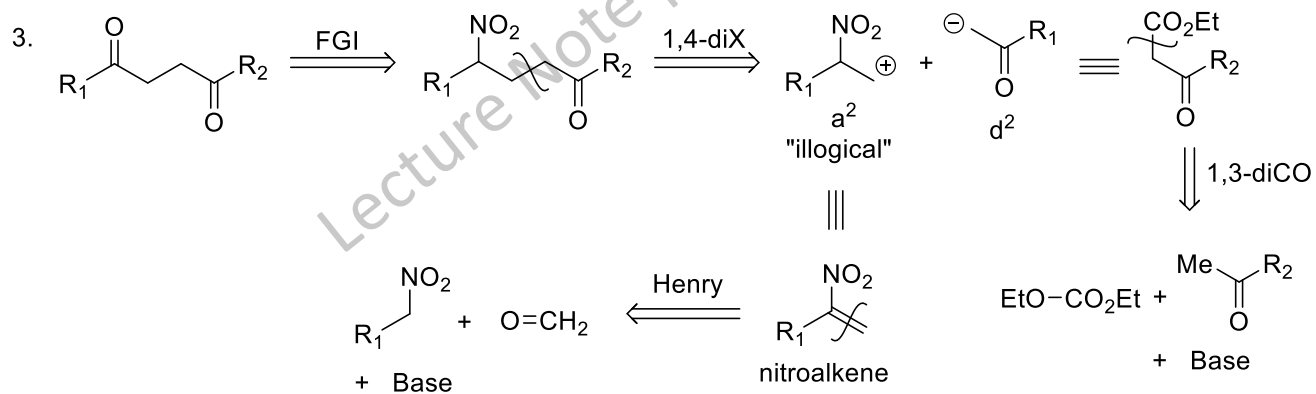
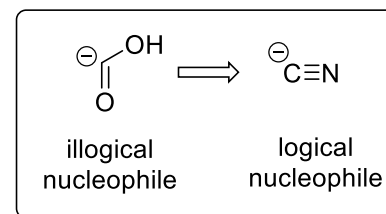
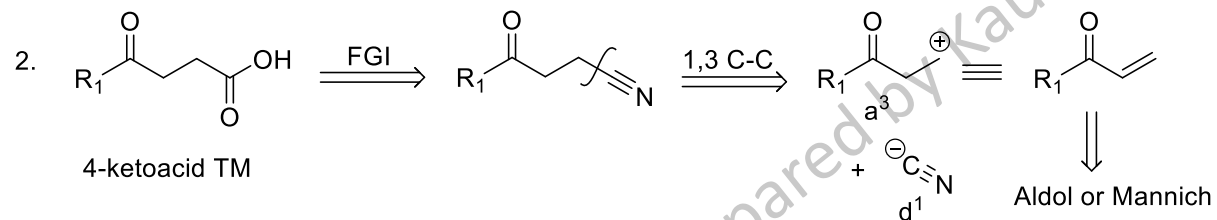
The Logic of Organic Synthesis: Analysis of bifunctional target molecules:

C] 1,4-bifunctional compounds:

The following is a summary of the retrosynthetic strategies one can commonly adopt when the target molecule contains two heteroatom-based functional groups placed at an 1,4-relation. These target molecules are dissonant systems, so umpolung strategy will be necessary.



1. Nitroalkane anions are excellent Michael donors
 2. Demasking nitro to carbonyl by McMurry reaction, TiCl_3 , $\text{H}_3\text{O}^{\oplus}$

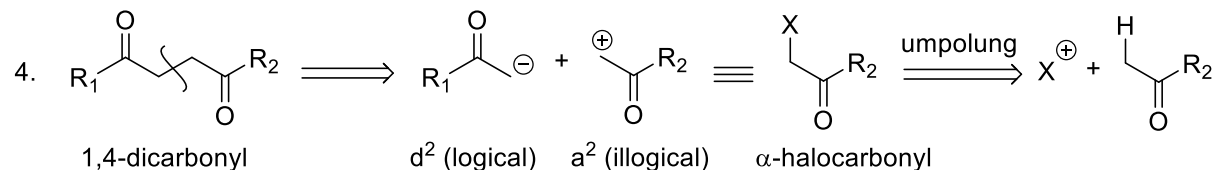


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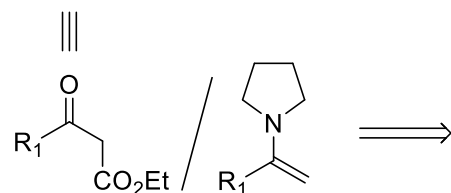
The Logic of Organic Synthesis: Analysis of bifunctional target molecules:

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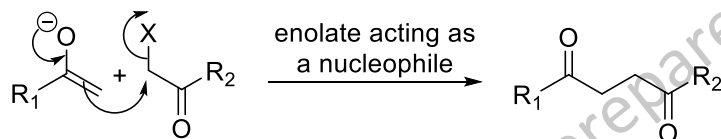


halogenation at the α -position may include regioselectivity issue

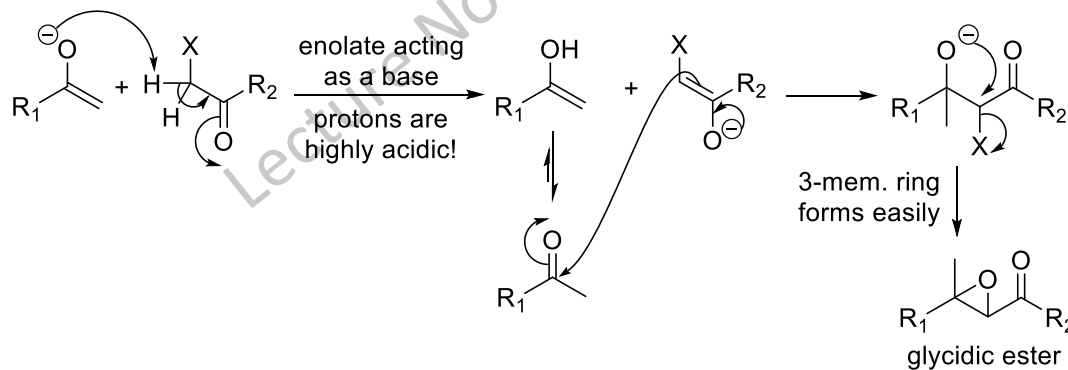


specific enol equivalent is needed for the d^2 synthon otherwise Darzen's reaction would take over...

desired outcome:



possible side reaction:

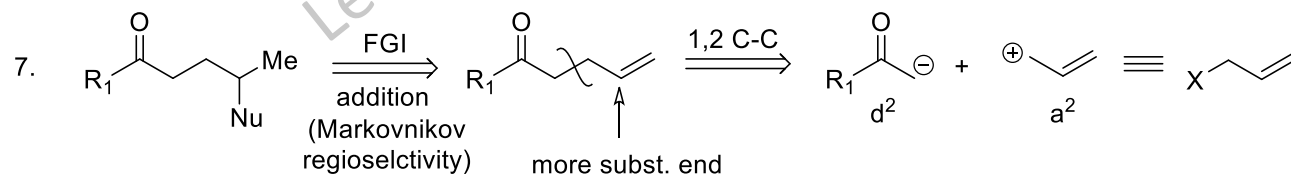
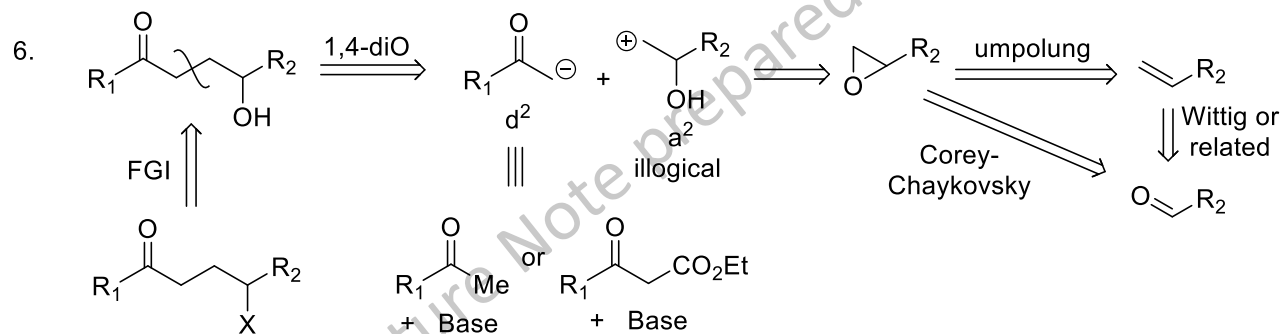
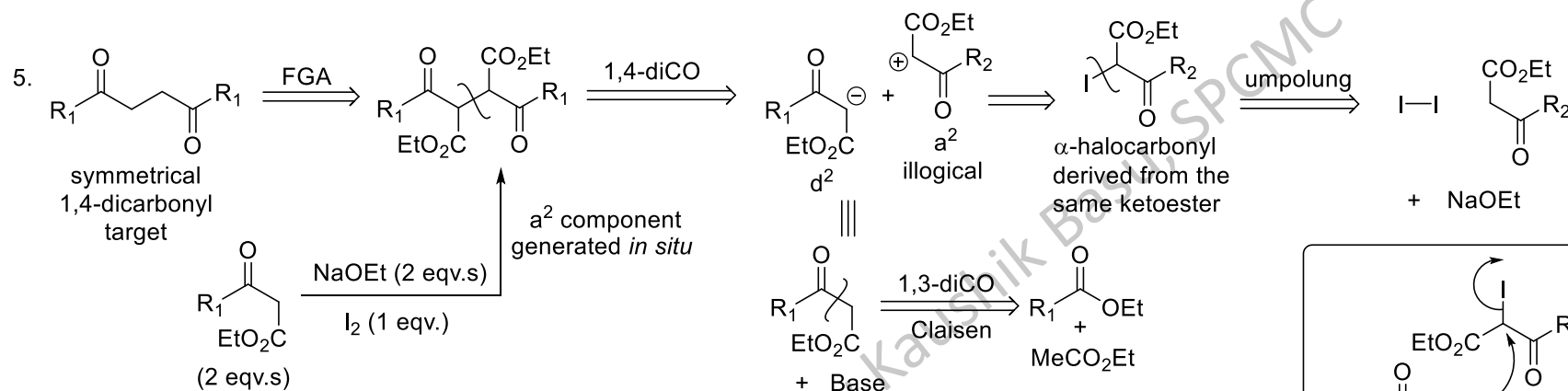


to minimise this, we must reduce the basicity of the enolate; we need specific enol equivalents

The Logic of Organic Synthesis: Analysis of bifunctional target molecules:

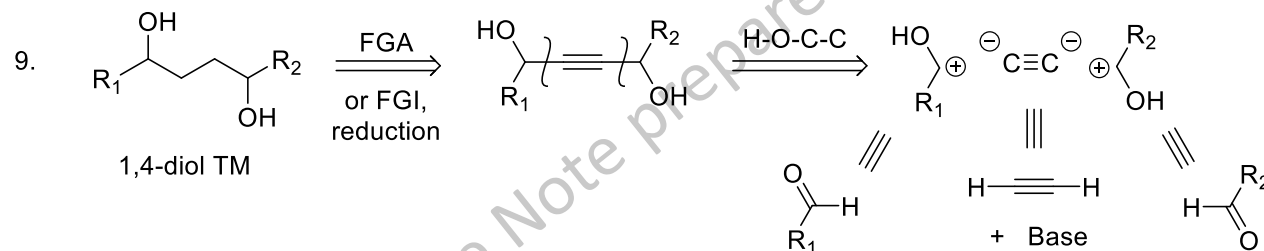
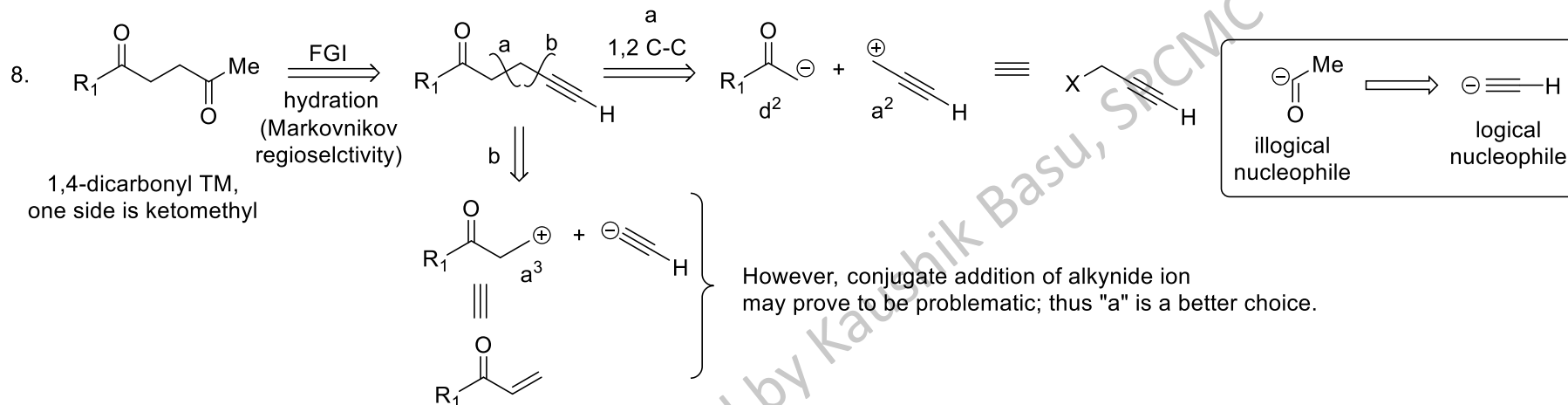
C] 1,4-bifunctional compounds (contd.):

The following is a summary of the retrosynthetic strategies one can commonly adopt when the target molecule contains two heteroatom-based functional groups placed at an 1,4-relation. These target molecules are dissonant systems, so umpolung strategy will be necessary.

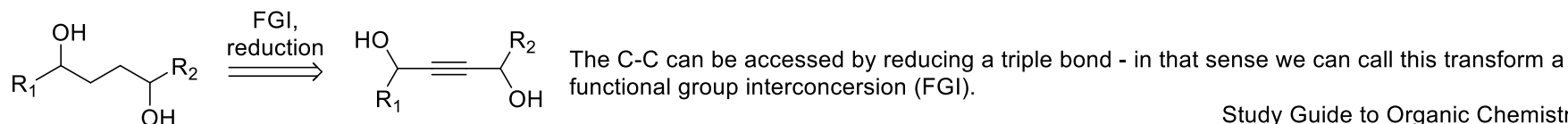
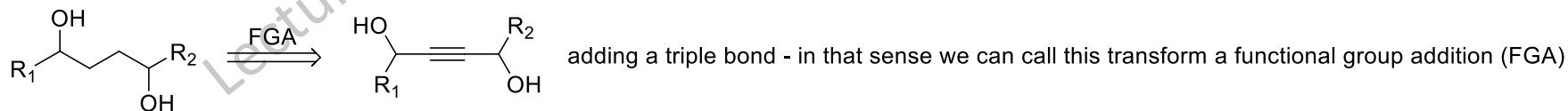


The Logic of Organic Synthesis: Analysis of bifunctional target molecules:
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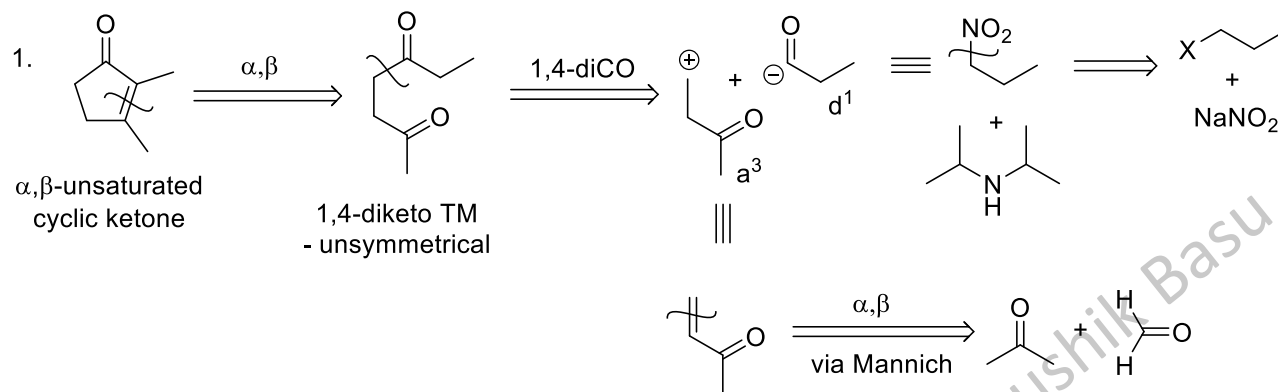
Note:



The Logic of Organic Synthesis: Analysis of bifunctional target molecules:

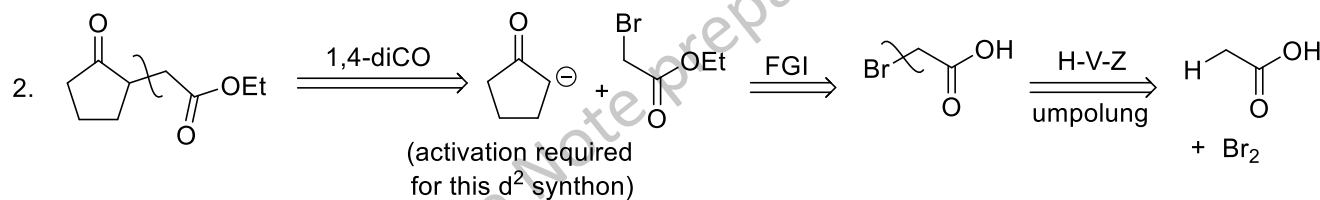
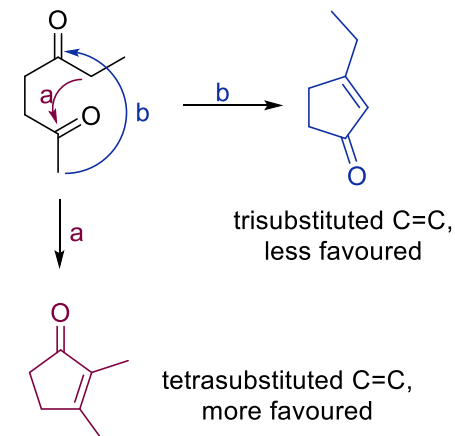
C] 1,4-bifunctional compounds (contd.):

Let us now consider a few examples:

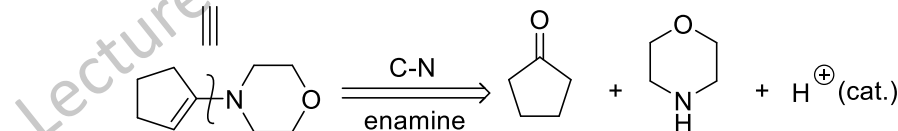
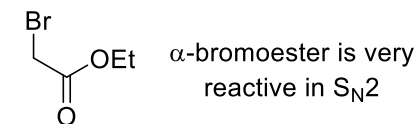


nitroalkane as acyl anion eqv.

condensation regioselectivity guided by formation of the more substituted C=C



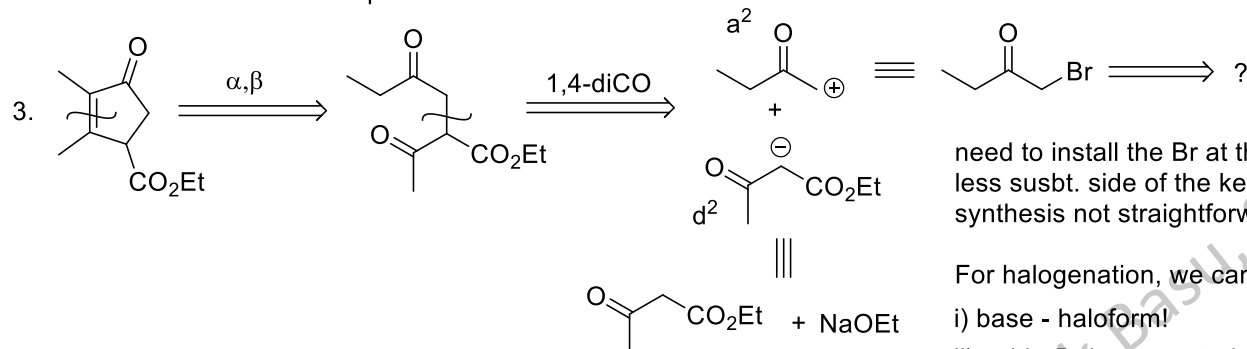
enamines react particularly well with $\text{S}_{\text{N}}2$ reactive electrophiles



The Logic of Organic Synthesis: Analysis of bifunctional target molecules:

C] 1,4-bifunctional compounds (contd.):

Let us now consider a few examples:



condensation regioselectivity
guided by formation of the
more substituted C=C

need to install the Br at the
less subst. side of the ketone,
synthesis not straightforward.

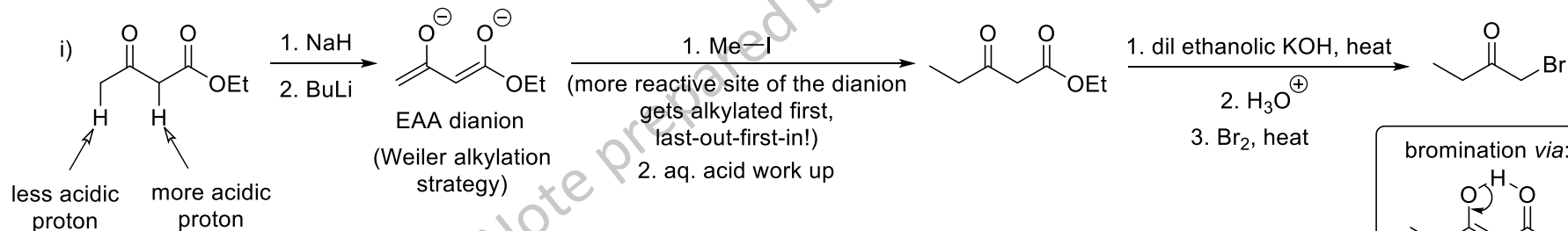
For halogenation, we can't use

i) base - haloform!

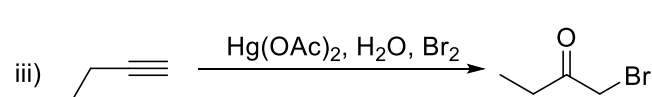
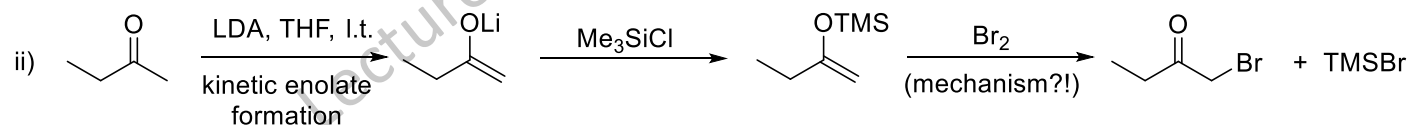
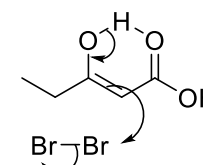
ii) acid - Br incorporated on the more substituted side!

How to solve this regioselectivity issue?

* Solutions to the regioselectivity problem:



bromination via:

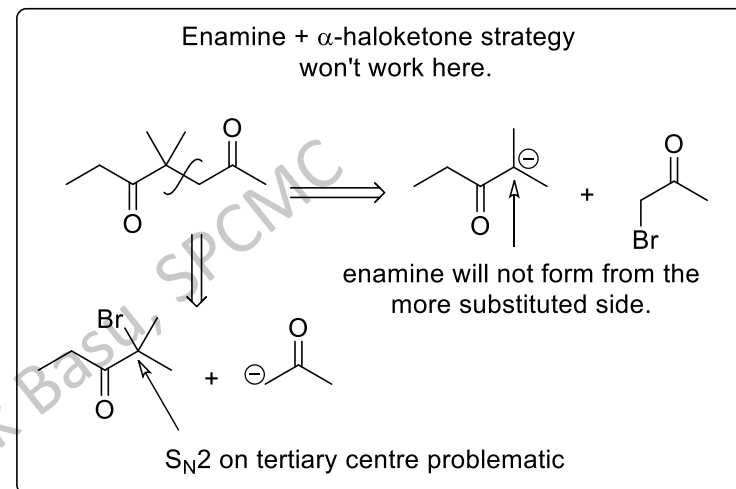
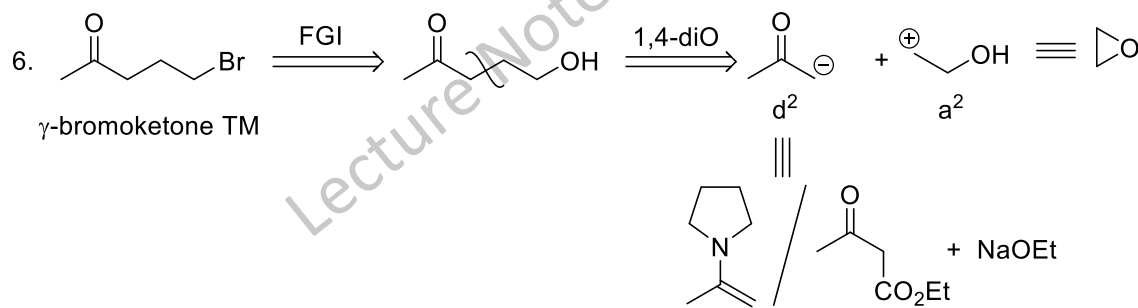
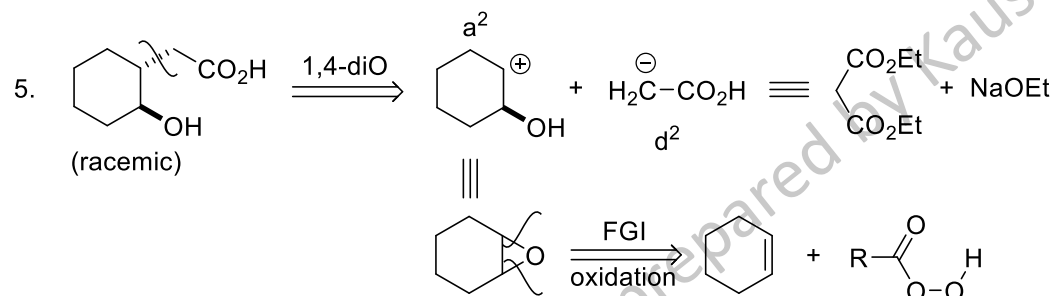
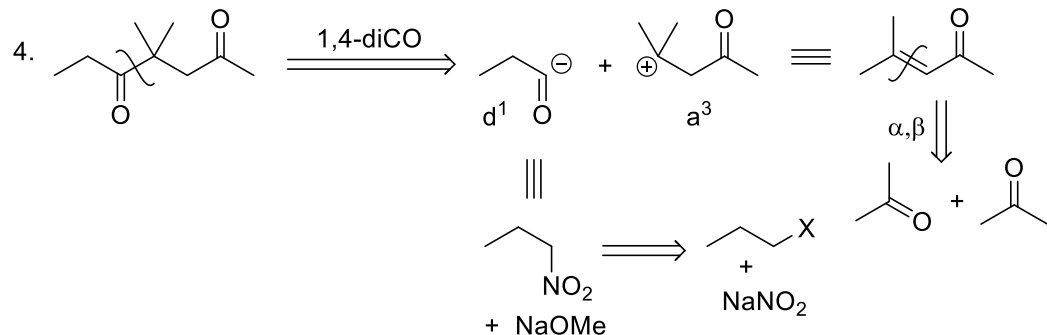


(This one is most interesting of the lot and
as expected, has the most intricate mechanism.
Try it; start just as you would for a Hg(II)-catalysed
hydration of alkyne and then proceed from there.)

The Logic of Organic Synthesis: Analysis of bifunctional target molecules:

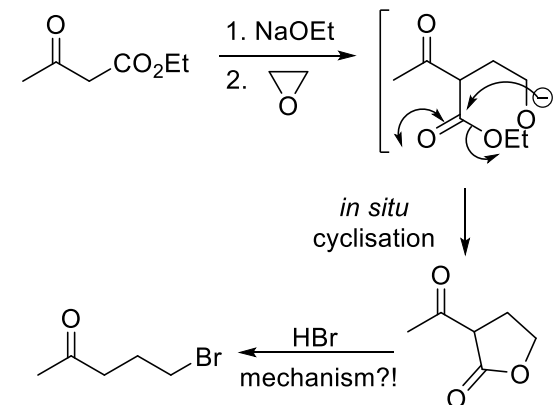
C] 1,4-bifunctional compounds (contd.):

Let us now consider a few examples:



epoxide ring-opening is in *trans*-orientation

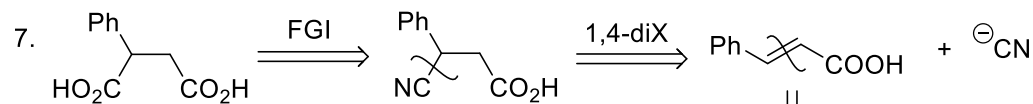
Using EAA route we have the following observation:



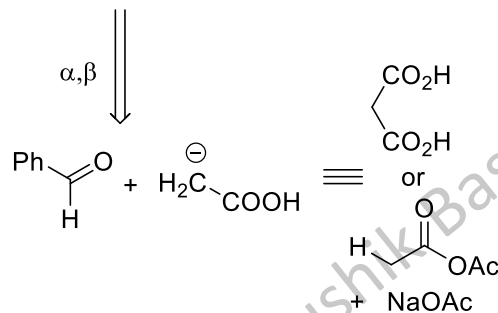
The Logic of Organic Synthesis: Analysis of bifunctional target molecules:

C] 1,4-bifunctional compounds (contd.):

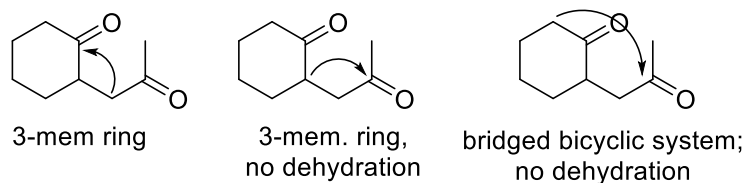
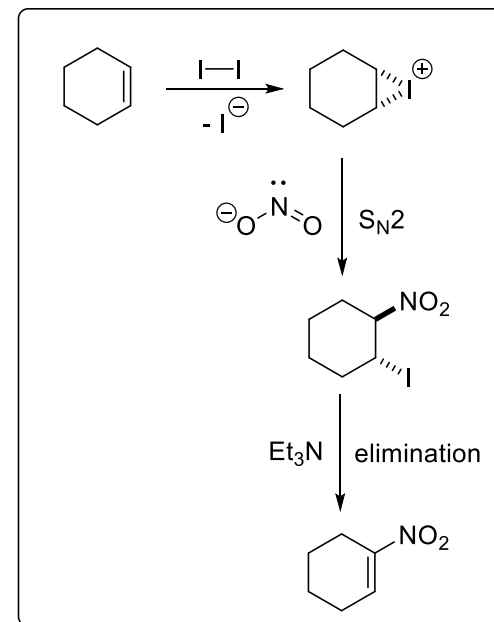
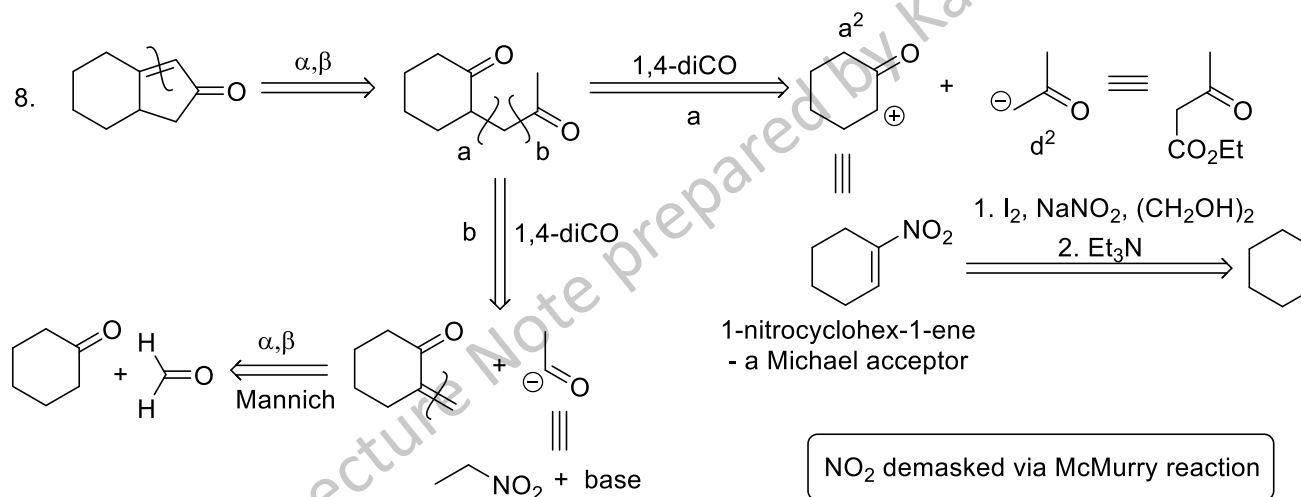
Let us now consider a few examples:



conjugate addition of cyanide is inevitable as it cannot add to COOH.



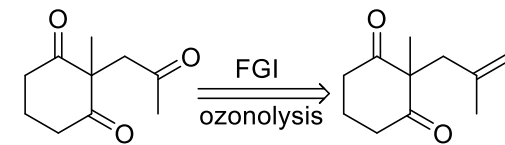
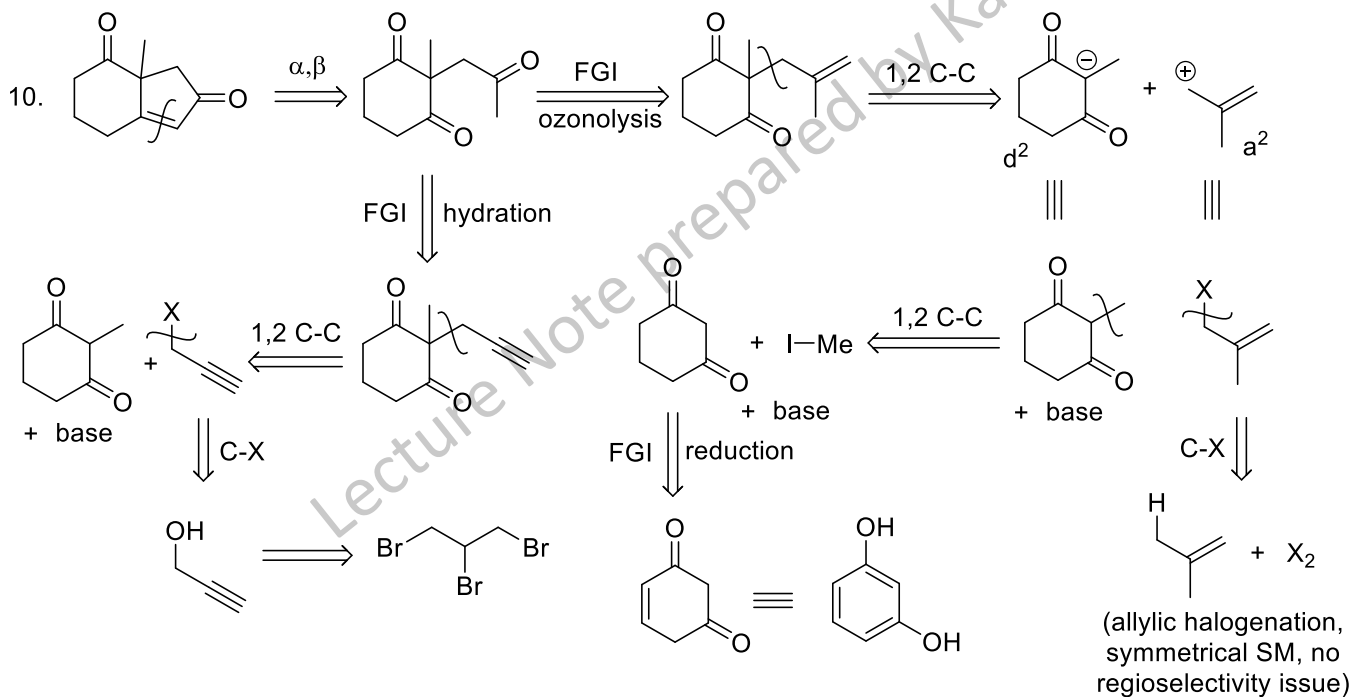
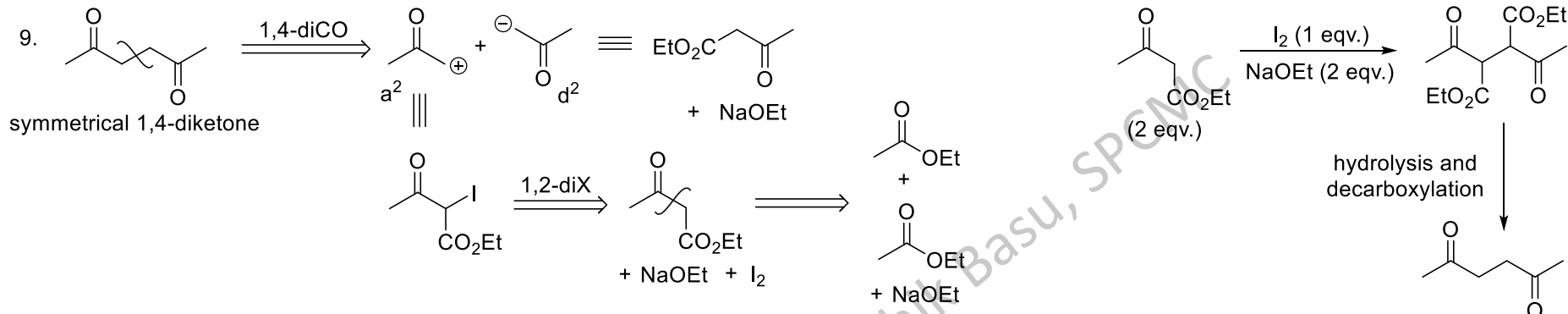
Use Perkin or Knoevenagel to access the aromatic α, β -unsaturated acid



The Logic of Organic Synthesis: Analysis of bifunctional target molecules:

C] 1,4-bifunctional compounds (contd.):

Let us now consider a few examples:



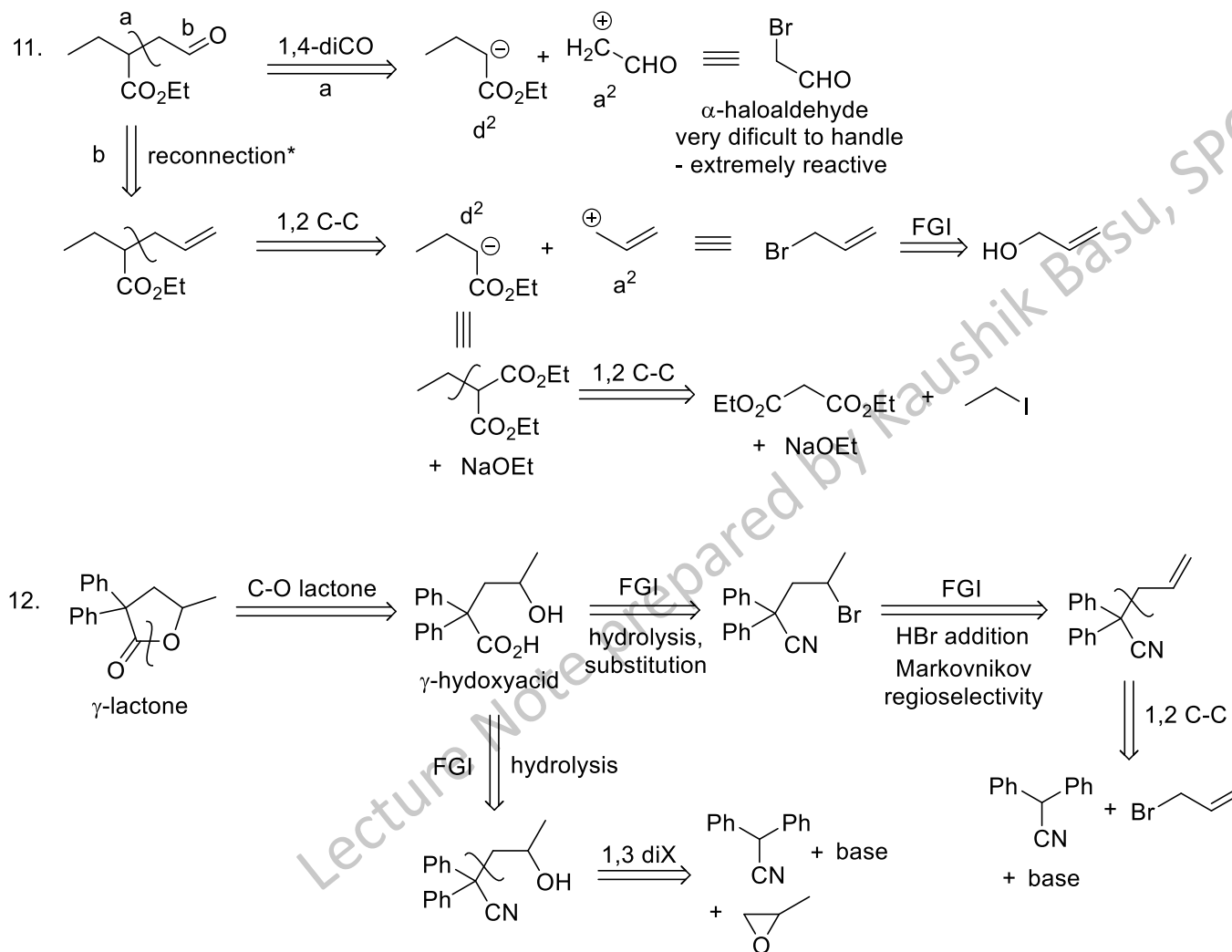
One of the strategies used here is quite unique - the ozonolysis of a C=C to install a C=O group. In terms of retrosynthesis, this implies replacing a =O with =CH₂. That's not a disconnection *per se*. When this analysis is carried out to approach the dicarbonyl target, we are in fact using the *reverse of a disconnection* - we are joining up a bond in the revised target which will be broken during the synthesis.

This is called the *strategy of reconnection*.

The Logic of Organic Synthesis: Analysis of bifunctional target molecules:

C] 1,4-bifunctional compounds (contd.):

Oxidative cleavage of a C=C provides a useful route to 1,4-dicarbonyl targets. Here's another example of this concept at work:

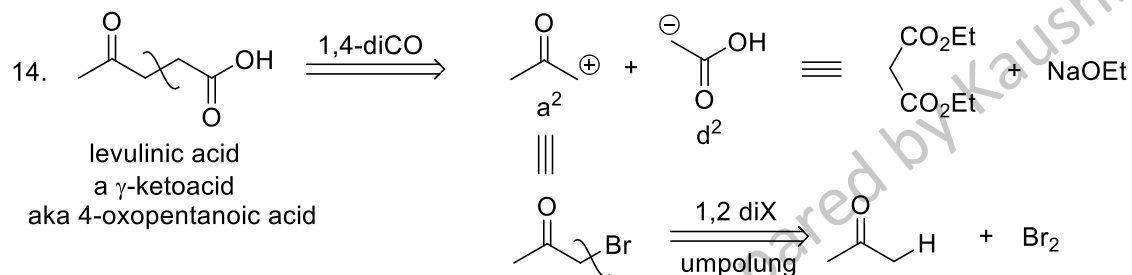
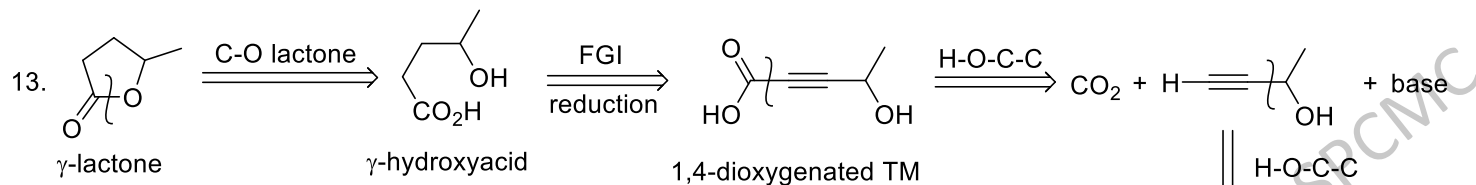


* to stop the ozonolysis of the C=C at the aldehyde stage, we need a reductive work-up, use dimethyl sulfide

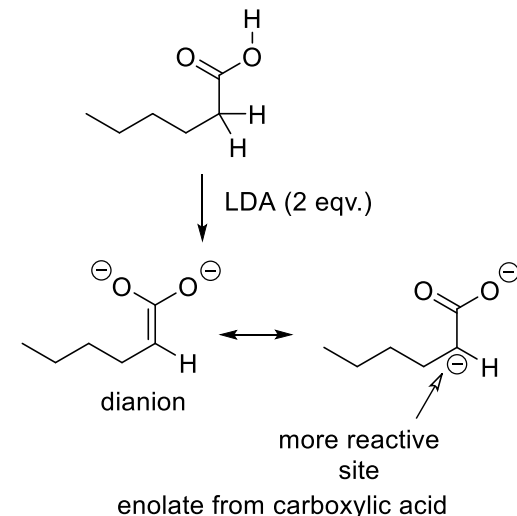
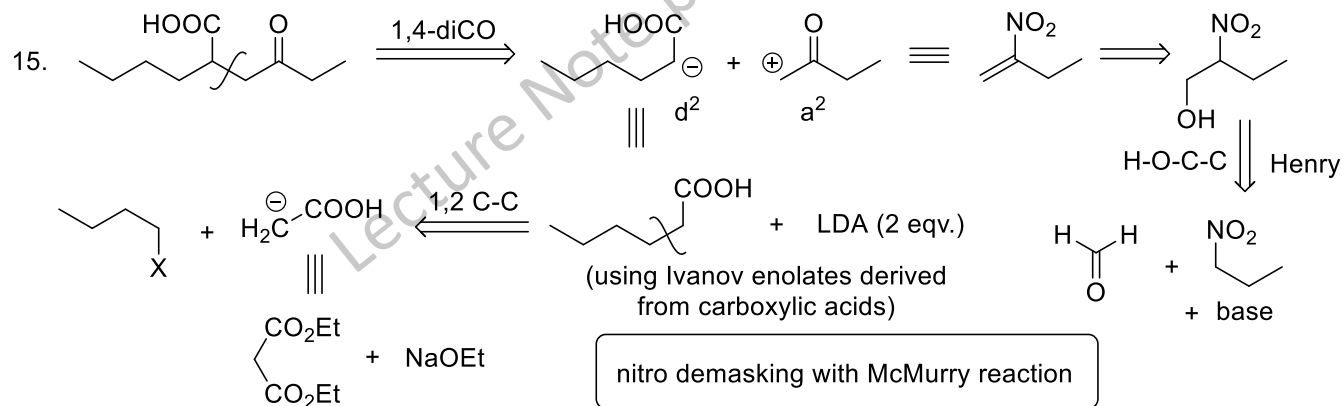
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C] 1,4-bifunctional compounds (contd.):

Let us now consider a few examples:

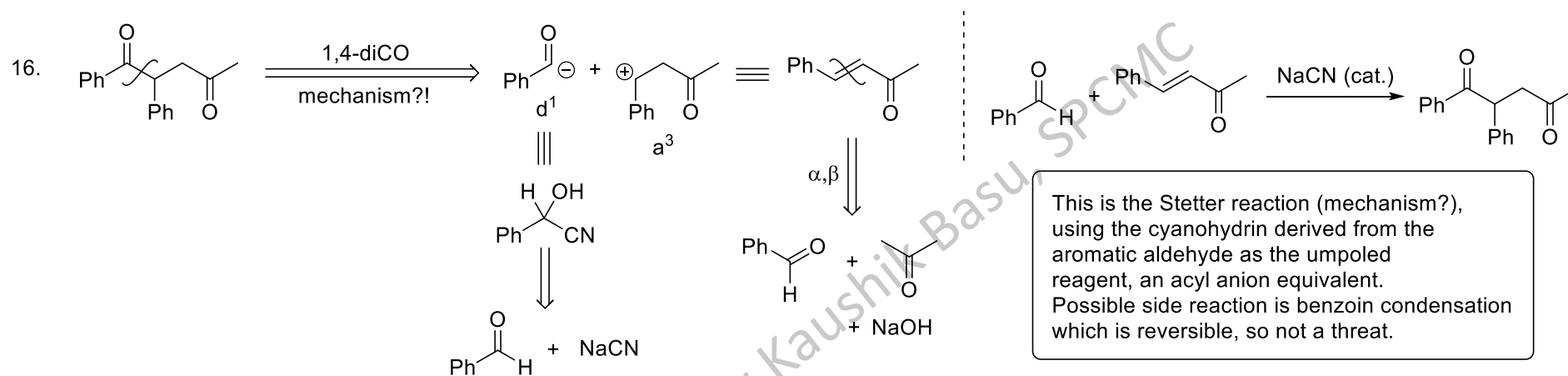


bromination in AcOH to access the α -bromoacetone

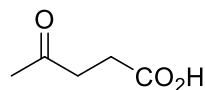


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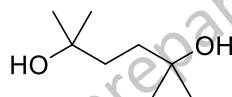
C] 1,4-bifunctional compounds (contd.):

 And finally, let us revisit a 1,4-dicarbonyl target once more. Again we use the d^1+a^3 combination, but this time, our acyl anion equivalent is different from the one derived from nitroalkanes:


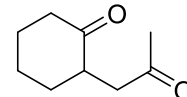
Try these yourself:



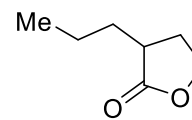
[C.1]

 (two separate methods,
other than the one shown)


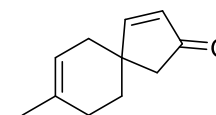
[C.2]



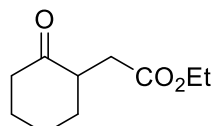
[C.3]

 (other than the
methods shown)


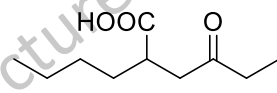
[C.4]



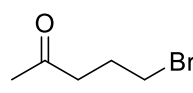
[C.5]



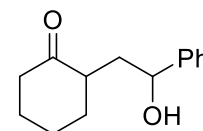
[C.6]



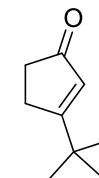
[C.7]

 (other than the
method shown)


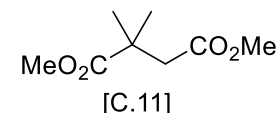
[C.8]



[C.9]



[C.10]



[C.11]